Safety Data Sheet

Denatured alcohol (95% Ethanol)

Product Code: SO-AL0020

Department: solvents

C.A.S.: 64-17-5, 67-56-1, 141-78-6, 7732-18-5



Section: 1 Identification

Product Name: Denatured Ethyl Alcohol DA-2A (Anhydrous)

Synonyms: 2A Alcohol or solvent Chemical Family: Alcohol / Esters.

Application: General purpose organic solvent, printing inks, protective and

decorative coatings, resins.

24-Hour Emergency Telephone Number (CANUTEC): (613) 996-6666

Section: 2 Hazard Identification

HGS Label Elements



Signal Word

Danger

Ingestion:

GHS Classification

Flammable liquids Cat.2 Acute toxicity - Oral Cat.4 Acute toxicity - Dermal Cat.3

Acute toxicity - Inhalation (Dusts/Mists) Cat.3

Skin corrosion/irritation Cat.2

Serious eye damage/eye irritation Cat.2A

Specific target organ toxicity (single exposure) Cat.1

Hazard Statements

Highly flammable liquid and vapor May cause respiratory irritation Causes damage to organs Toxic in contact with skin Toxic if inhaled Harmful if swallowed Causes skin irritation Causes serious eye irritation

Precautionary Statements

Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking

Keep container tightly closed

Ground/bond container and receiving equipment

Use explosion-proof electrical/ventilating/lighting/equipment

Use only non-sparking tools

Take precautionary measures against static discharge

Do not breathe dust/fume/gas/mist/vapors/spray

Wash face, hands and any exposed skin thoroughly after handling

Wear protective gloves/eye protection/face protection IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing

IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Rinse skin with water/shower

IF INHALED: Remove person to fresh air and keep comfortable for breathing

IF SWALLOWED: Call a POISON CENTER or doctor if you feel unwell Rinse mouth

In case of fire: Use dry sand, dry chemical or alcohol-resistant foam to extinguish

Section: 3 Composition / Information on Ingredients

Chemical name % weight (W/W) CAS no. Synonyms 80 - 90%

64-17-5 Ethanol Methanol 67-56-1 ethyl Acetate 141-78-6 water 7732-18-5

Section: 4 First-Aid Measures

Eye Contact: Immediately flush eyes with copious quantities of water for at least 20 minutes holding

lids apart to ensure flushing of the entire surface. Seek immediate medical attention. Skin Contact: In case of contact, immediately flush skin with plenty of water for at least 15 minutes.

Get medical attention. Remove contaminated clothing and launder before reuse. If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is Inhalation:

10 - 20%

0 - 10%

0 - 10%

difficult, give oxygen. If the heart has stopped, trained personnel should begin

cardiopulmonary resuscitation (CPR) immediately. Get medical attention immediately.

Do NOT induce vomiting. Never give anything by mouth to an unconscious or

convulsing person. Seek immediate medical attention. If vomiting occurs

spontaneously, keep head below hips to prevent aspiration of liquid into the lungs. Notes to Physician: Treatment based on sound judgment of physician and individual reactions of patient.

This product contains methanol, a toxic substance having produced blindness and other serious effects on vision, as well as death. However, this product also contains

the accepted antidote, ethanol.

Section: 5 Fire-Fighting Measures

Suitable Extinguishing Media

Specific hazards arising from the substance or mixture

Hazardous combustion products Special protective equipment and precautions for fire-fighters Use extinguishing measures that are appropriate to local circumstances and the surrounding environment.

Vapors from this product and may travel or be moved by air currents and ignited by pilot lights, other flames, smoking, sparks, heaters, electrical equipment, static discharges or other ignition sources at locations distant from product handling point. Use water spray to cool fire-exposed containers and structures. Use water spray to disperse vapors; reignition is possible.

Carbon monoxide. Carbon dioxide. Formaldehyde.

Firefighters should wear self-contained breathing apparatus and full firefighting turnout gear. Use personal protection equipment.

Section: 6 Accidental Release Measures

Personal Precautionary Measures: Environmental Precautionary Measures:

Procedure for Clean Up:

Wear appropriate protective equipment.

Prevent entry into sewers or streams, dike if needed. Consult local authorities.

Isolate hazard area and restrict access. Stop leak only if safe to do so. Remove ignition sources and work with non-sparking tools. Small spills: soak up with absorbent material and scoop into containers. Large spills: prevent contamination of waterways. Dike and pump into suitable containers. Clean up residual with absorbent material, place in appropriate container and flush with water.

Section: 7 Handling And Storage

Handling:

Storage:

For industrial use only. Handle and open containers with care. Avoid contact with eyes, skin and clothing. Do not ingest. Avoid inhalation of chemical. DO NOT handle or store near an open flame, heat, or other sources of ignition. Fixed equipment as well as transfer containers and equipment should be grounded to prevent accumulation of static charge. DO NOT pressurize, cut, heat, or weld containers. Empty containers may contain hazardous product residues. Keep the containers closed when not in use. Protect against physical damage. Use appropriate personnel protective equipment.

Store in a cool, dry, well ventilated area, away from heat and ignition sources. Place away from incompatible materials. Store in accordance with good industrial practices.

Section: 8 Exposure Control/Personal Protection

Chemical name	Alberta OEL OEL	Brit. Columbia OEL	Ontario	Quebec OEL	Exposure Limit ACGIH.	Immediately Dangerous to life or Health -IDLH
Ethanol 64-17-5	TWA: 1000 ppm TWA: 1880 mg/m3	STEL:1000 ppm	STEL:1000 ppm	TWA: 1000 ppm TWA: 1880 mg/m3	1000 ppm STE	L3300 ppm
Methanol 67-56-1	TWA: 200 ppm TWA: 262 mg/m3 STEL: 250 ppm STEL: 328 mg/m3 skin	TWA: 200 ppm STEL:250 ppm Skin	TWA: 200 ppm STEL:250 ppm skin	TWA: 200 ppm TWA: 262 mg/m3 STEL: 250 ppm STEL: 328 mg/m3 skin	250 ppm STEL 200 ppm TLV-TWA	6000 ppm
Ethyl Acetate 141-78-6	TWA: 400 ppm TWA: 1440 mg/m3	TWA: 150 ppm	TWA: 400 ppm	TWA: 400 ppm TWA: 1440 mg/m3	TWA: 400 ppm TLV-TWA	2000ppm
water	not available 7732-18-5	not available	not available	not available	not available	not available

Consult local authorities for recommended exposure limits

Appropriate engineering controls

Engineering controls

Use process enclosure, local exhaust ventilation, or other engineering controls to keep

airborne levels below recommended exposure limits. Use explosion proof equipment.

Individual protection measures, such as personal protective equipment

Eye/face protection Chemical goggles; also wear a face shield if splashing hazard exists.

Hand protection Butyl rubber gloves. Neoprene gloves. Rubber gloves.

Skin and body protection Skin contact should be prevented through the use of suitable protective clothing, gloves

and footwear, selected for conditions of use and exposure potential. Consideration must be

given both to durability as well as permeation resistance.

Respiratory protection Up to 1000 ppm, an approved organic vapor cartridge respirator can be used. For

concentrations above 1000 ppm, an air-supplying respirator is recommended. The user should consult a respirator guide, such as the Canadian Standards Association's guide

Z94.4- M1982.

General hygiene considerations Avoid contact with skin, eyes or clothing. Wear suitable gloves and eye/face protection. Do

not eat, drink or smoke when using this product. Remove and wash contaminated clothing and gloves, including the inside, before re-use. Regular cleaning of equipment, work area and clothing is recommended. Wash hands before breaks and immediately after handling

the product.

Section: 9 Physical and Chemical Properties

Physical State: Liquid. Colour: Colourless Odour: Alcohol Not Available. pH: Specific Gravity: 0.7889 **Boiling Point:** 75.6°C /168°F Freezing/Melting Point: Not Available. Vapour Pressure: Not Available. Vapour Density: Not Available. 100% % Volatile by Volume:

Evaporation Rate:

Solubility:

VOCs:

Viscosity:

Molecular Weight:

Other:

1.8

Completely soluble.

Not Available.

Not Available.

Not Available.

Not Available.

Section: 10 Stability And Reactivity

Chemical Stability: Stable.

Hazardous Polymerization: Will not occur.

Conditions to Avoid: Avoid excessive heat, open flames and all ignition sources.

Materials to Avoid: Oxidizing materials.

Hazardous Decomposition Products: Carbon monoxide. Carbon dioxide. Formaldehyde.

Additional Information: No additional remark.

Section: 11 Toxicological Information

Information on likely routes of exposure Inhalation

Eye contact

Skin contact

Ingestion

Information on toxicological effects Symptoms

High vapor concentrations are irritating to the eyes, nose, throat and lungs; may cause headaches and dizziness; may be anesthetic and may cause other central nervous system effects.

Causes moderate to severe irritation, experienced as discomfort or pain, excess blinking and tear production, with marked excess redness and swelling of the conjunctiva.

Causes mild skin irritation. May cause dermatitis, prolonged or repeated contact may cause skin sensitization. May be absorbed through the skin and contribute to the symptoms listed under ingestion.

A small amount of methanol (usually two or more ounces) can cause mental sluggishness, nausea and vomiting leading to severe illness, and may produce adverse effects on vision with possible blindness or death if treatment is not received. May cause headache, nausea, abdominal discomfort, vomiting, diarrhea, dizziness, drowsiness, faintness, lack of coordination and unconsciousness.

Repeated exposure by inhalation or absorption of methanol may cause systemic poisoning, brain disorders, impaired vision and blindness. Inhalation may worsen conditions such as emphysema or bronchitis. Repeated skin contact may cause dermal irritation, dryness and cracking. Effects of sub lethal doses may be nausea, headache, abdominal pain, vomiting and visual disturbances ranging from blurred vision to light sensitivity. Methanol is toxic by inhalation and ingestion. Inhalation of vapors may cause cyanosis, CNS effects, lethargy, loss of consciousness and death. The effects from inhalation may be delayed. Ingestion may cause malaise, CNS effects, discomfort, and death if not treated promptly. Ingestion of methanol has resulted in adverse effects (necrosis and hemorrhaging) in the brain. Medical conditions aggravated by exposure include: skin disorders and allergies, liver disorders and eye disease. Long term exposure to methanol has been associated with headaches,

Synergistic Products:

Potential for Accumulation:

giddiness, conjunctivitis, insomnia and impaired vision. Dermal absorption of significant amounts of methanol resulted in death in several animal species. Toxic effects in animals exposed to methanol by inhalation include eye irritation, blindness and nasal discharge. Toxic effects observed in animals exposed to methanol by ingestion include CNS effects, gastrointestinal effects, anesthetic effects, damage to the optic nerve and acidosis.

In animals, high concentrations of methanol can increase the toxicity of other chemicals, particularly liver toxins like carbon tetrachloride. Ethanol significantly reduces the toxicity of methanol because it competes for the same metabolic enzymes, and has been usd to treat methanol poisoning.

Methanol is readily absorbed into the body following inhalation and ingestion. Skin absorption may occur if the skin is broken or exposure is prolonged. Once absorbed, methanol is rapidly distributed to body tissues. A small amount is excreted unchanged in exhaled air and the urine. The rest is first metabolized to formaldehyde, which is then metabolized to formic acid and/or formate. The formic acid and formate are eventually converted to carbon dioxide and water. In humans, methanol clears from the body, after inhalation or oral exposure, with a half-life of 1 day or more for high doses (greater than 1000 mg/kg) or about 1.5-3 hours for low doses (less than 100 mg/kg or 76.5-230 ppm (100-300 mg/m3)). Long term repeated oral exposure to ethanol may result in the development of progressive liver injury with fibrosis. Synergistic Materials: Ethanol with carbon tetrachloride, chloroform, bromotrichloromethane, dimethylnitrosamine, thioacetamide, methanol with carbon tetrachloride.

Numerical measures of toxicity

Acute toxicity

The following values are calculated based on chapter 3.1 of the GHS document .

 ATEmix (oral)
 671.00 mg/kg

 ATEmix (dermal)
 318.00 mg/kg

 ATEmix (inhalation-dust/mist)
 0.53 mg/l

Chemical name	Oral LD50	Dermal LD50	Inhalation LC50
Ethanol 64-17-5	= 7060 mg/kg (Rat)	Not available	= 124.7 mg/L (Rat) 4 h
Methanol 67-56-1	= 6200 mg/kg (Rat)	= 15800 mg/kg (Rabbit) = 15840 mg/kg (Rabbit)	= 22500 ppm (Rat) 8 h = 64000 ppm (Rat) 4 h
Ethyl acetate 141-78-6	= 5620 mg/kg (Rat)	> 18000 mg/kg (Rabbit) > 20 mL/kg (Rabbit)	= 4000 ppm (Rat) 4 h
water 7732-18-5	> 90 mL/kg (Rat)	Not available	Not available

Delayed and immediate effects as well as chronic effects from short and long-term exposure

Skin corrosion/irritation

Serious eye damage/eye irritation

Respiratory or skin sensitization Germ cell mutagenicity

Carcinogenicity

Causes mild skin irritation. May cause dermatitis, prolonged or repeated contact may cause skin sensitization. May be absorbed through the skin and contribute to the symptoms listed under ingestion.

Causes moderate to severe irritation, experienced as discomfort or pain, excess blinking and tear production, with marked excess redness and swelling of the conjunctiva.

No information available. No information available.

Ethanol possesses properties that indicate a carcinogenicity hazard for human health but these are manifest only at doses associated with consumption of alcoholic beverages. In the context of an industrial chemical, these hazards do not warrant concern as these are not likely to result from the manufacture and use of ethanol and ethanol containing products.

The table below indicates whether each agency has listed any ingredient as a carcinogen.

Chemical name ACGIH AIRCNTPOSHA

Ethanol A3 Group 1knownX

64-17-5

Methanol Not available Not availableNot availableNot

available

67-56-1

Ethyl acetate Not available Not availableNot availableNot

available

141-78-6

Water Not available Not availableNot availableNot

available

7732-18-5

Legend

ACGIH A3 - Animal Carcinogen

(American Conference of Governmental Industrial Hygienists)

IARC Group 1 - Carcinogenic to Humans

(International Agency for Research on Cancer)

NTP Known - Known Carcinogen

(National Toxicology Program)

OSHA X - Present

(Occupational Safety and Health Administration of the US Department of Labor)

Reproductive toxicity Methanol is reported to cause birth defects in rats exposed to 20 000 ppm. In

experimental animals, methanol is fetotoxic, teratogenic and has produced significant behavioral abnormalities in offspring at dose levels not producing maternal toxic effects. Behavioral abnormalities were observed in the offspring of rats given drinking water containing 2% methanol. Methanol has produced mutagenic effects (somatic cells) in experimental animals. Contains Ethanol, which may cause birth defects or other adverse effects on pregnancy. Risk of

effects depends on duration and level of

exposure.

- repeated exposure

Specific target organ systemic toxicity

- single exposure Based on the classification criteria of the Globally Harmonized System as

adopted in the country or region with which this safety data sheet complies, this product has been determined to cause systemic target organ toxicity from acute

exposure. (STOT SE). Causes damage to organs.

No information available.

Aspiration hazard No information available.

Section: 12 Ecological Information

Ecotoxicological Information:

Ingredients Ecotoxicity Fish Species Data Crustacea Ecotoxicity Freshwater

Algae Data

Not Available.

Not Available

Ethanol LC50 96 h (Oncorhynchus mykiss)

12.0-16.0 ml/L static

LC50 96 h (Pimephales promelas)

>100 mg/L static

LC50 96 h (Pimephales promelas) 13400-15100 mg/L flow-through LC50 96 h (Oncorhynchus mykiss)

12.0-16.0 ml/L static

=2mg/L (48h, Daphnia magna)

LC50: 9268 - 14221mg/L

EC50: =10800mg/L (24h

Daphnia magna) EC50:

(48h. Daphnia magna)

Methanol LC50 (Oncorhynchus mykiss)

13200 mg/L

LC50 (Pimephales promelas)

28100 mg/L (96 hrs)

LC50 (Lepomis macrochirus)

15400 mg/L (96 hrs)

Ethyl Acetate 220 - 250 mg/L LC50

(Pimephales promelas) 96 h flow-through 352 -500 mg/L LC50 (Oncorhynchus mykiss) Not Available

Not Available

EC50: =560mg/L (48h, Daphnia magna)

Water Not Available Not Available Not Available.

Other Information: No additional remark.

Section: 13 Disposal Considerations

Disposal of Waste Method: Disposal of all wastes must be done in accordance with municipal, provincial and

federal regulations.

Contaminated Packaging: Empty containers should be recycled or disposed of through an approved waste

management facility.

Section: 14 Transport Information

DOT (U.S.):

DOT Shipping Name: ALCOHOLS, FLAMMABLE, TOXIC, N.O.S. (ETHANOL)

DOT Hazardous Class: 3 (6.1)
DOT UN Number: UN1986
DOT Packing Group: II

DOT Reportable Quantity (lbs): Not Available.

Note: No additional remark.

Marine Pollutant: No.

TDG (Canada):

TDG Shipping Name: ALCOHOLS, FLAMMABLE, TOXIC, N.O.S. (ETHANOL)

Hazard Class: 3 (6.1)
UN Number: UN1986
Packing Group: II

Note: No additional remark.

Marine Pollutant: No.

Section: 15 Regulatory Information

U.S. TSCA Inventory Status: All components of this product are either on the Toxic Substances Control Act (TSCA) Inventory

List or exempt.

Canadian DSL Inventory Status: All components of this product are either on the Domestic Substances List (DSL), the Non-

Domestic Substances List (NDSL) or exempt.

U.S. Regulatory Rules

Ingredients CERCLA/SARA Section 302: SARA (311, 312) Hazard Class: CERCLA/SARA –

Section 313:

Ethanol Not Listed. Not Listed. Not Listed. Not Listed. Listed Methanol Listed Ethyl Acetate Not Listed. Listed Not Listed. Water Not Listed. Not Listed. Not Listed.

California Proposition 65: Listed.
MA Right to Know List: Listed.
New Jersey Right-to-Know List: Listed.
Pennsylvania Right to Know List: Listed.

Section: 16 Other Information

Reference Manufacturer's material safety data sheet.

Prepared by Kama pigments

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